



1 **Nighttime O(¹D) distributions in the mesopause region derived from** 2 **SABER data**

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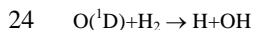
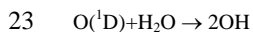
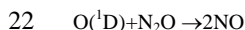
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6 **Abstract.** In this study, the new source of O(¹D) in the mesopause region due to the process OH($v \geq 5$) + O(³P) →
7 OH($0 \leq v' \leq v-5$) + O(¹D) is applied to SABER data to estimate the nighttime O(¹D) distributions for the years 2003-2005. It is
8 found that O(¹D) evolutions in these years are very similar to each other. Depending on the month, monthly averaged O(¹D)
9 distributions demonstrate from 2 to 4 maxima with the values up to 340 cm⁻³ which are localized in height (at ~92-96 km)
10 and latitude (at ~20-40°S,N and ~60-80°S,N). Annually averaged distributions in 2003-2005 have a one weak maximum at
11 ~93 km and ~65°S with the values of 150-160 cm⁻³ and 3 pronounced maxima (with the values up to 230 cm⁻³) at ~95 km
12 and ~35°S, at ~94 km and ~40°N, at ~93 km and ~65-75°N correspondingly. In general, there is slightly more O(¹D) in the
13 northern hemisphere than in the southern hemisphere. The obtained results are useful data set for subsequent estimation of
14 nighttime O(¹D) influence on chemistry of the mesopause region.

15 **1 Introduction**

16 Daytime O(¹D) is considered to be one of the important chemical minor species of the stratosphere, mesosphere and
17 thermosphere, as it plays a significant role in the chemistry, and the radiative and thermal balance of this region (Brasseur &
18 Solomon, 2005). First of all, formed by photolysis of O₂ and O₃, O(¹D) is a mediator involved in the transformation of
19 absorbed solar radiation energy into the heating of this region and, in particular, excitation of N₂(v) and CO₂(v) (Harris &
20 Adams, 1983; Panka et al., 2017). Also, O(¹D) atoms participate in the reactions of destruction of long-lived greenhouse
21 gases (Baasandorj et al., 2012), CH₄ oxidation, and HO_x and NO_x production, for example:



27 Moreover, the red line emission from O(¹D) atoms is one of the most important airglow phenomenon which are used as a
28 diagnostic of the ionosphere, for example, to monitor the electron density and neutral winds in the F region (Shepherd et al.,
29 2019). Therefore, many papers and experimental campaigns are devoted to measurements of features of O₃ photolysis to
30 O(¹D) (Taniguchi et al., 2003; Hofzumahaus et al., 2004).

31 Until recently, it was believed that the above mentioned processes stop at night as constant source of O(¹D) is absent while
32 the life time of the component is extremely low (less than 1 s). In principle, O(¹D) can be generated in sprite halos but for a
33 short duration of 1 ms (Hiraki et al., 2004). Recently, Sharma et al. (2015) and Kalogerakis et al. (2016) basing on laboratory
34 experiments proposed that O(¹D) could be produced in the mesopause region via process OH($v \geq 5$) + O(³P) → OH($0 \leq v' \leq v-5$)
35 + O(¹D), that is multiquantum quenching of high excited states of OH by collisions with atomic oxygen in ground state.

36 Last year, Kalogerakis (2019) showed that a new model of O₂ A-band, that takes this process into account, describes well
37 (qualitatively and quantitatively) the results of early nighttime rocket measurements of volume emission rate profiles of this
38 airglow. Thus, he proved that the process OH($v \geq 5$) + O(³P) → OH($0 \leq v' \leq v-5$) + O(¹D) really took place in nighttime
39 mesopause, and the produced O(¹D) distributions can be evaluated from available data.



40 In this study, the new source of $O(^1D)$ in the mesopause region is applied to SABER data to estimate the $O(^1D)$ nighttime
 41 distributions for the years 2003-2005.

42 2 $O(^1D)$ derivation from SABER Data

43 All processes used for $O(^1D)$ determination are summarized in Table 1. Here, we apply the new $OH(v)$ model of Fytterer et
 44 al. (2019). Their “best-fit model” includes all commonly used production and loss processes of $OH(v)$ (see Table 1), but
 45 some parameters of the model, in particular, branching ratios of quenching $OH(v)+O_2$ and rate coefficients of $OH(v \geq 5) +$
 46 $O(^3P) \rightarrow OH(0 \leq v' \leq v-5) + O(^1D)$ were adjusted with the use of volume emission rate profiles in four different wavelengths
 47 measured by SABER and SCIAMACHY.

48 Due to low values of chemical lifetimes (less than 1 s), $O(^1D)$ can be considered in chemical equilibrium:

$$49 \quad O(^1D) = \frac{k_9 \cdot OH(9) \cdot O(^3P) + k_{10} \cdot OH(8) \cdot O(^3P) + k_{11} \cdot OH(7) \cdot O(^3P) + k_{12} \cdot OH(6) \cdot O(^3P) + k_{13} \cdot OH(5) \cdot O(^3P)}{k_{14} + k_{15} \cdot O_2 + k_{16} \cdot N_2} \quad (2)$$

50 Thus, to calculate local value of $O(^1D)$ we should specify the local concentrations of $OH(v=5-9)$ and $O(^3P)$. The mentioned
 51 model lets us to derive the $OH(v)$ concentrations as the functions of the $OH(v)$ source due to the reaction (1) ($P_{OH} = k_1 \cdot H \cdot$
 52 O_3), air concentration (M), temperature (T), and $O(^3P)$ concentration:

$$53 \quad OH(v) = F_v(P_{OH}, M, T, O(^3P)) \quad (3)$$

54 To determine $O(^3P)$ and P_{OH} , we use known (e.g., Mlynczak et al., 2013, 2018) approach for $O(^3P)$ derivation from the
 55 simultaneous SABER measurements of volume emission rate of (9-7) and (8-6) OH transitions ($VER_{2\mu m}$), O_3 (9.6 μm), and
 56 temperature (T). The approach employs the chemical equilibrium condition for nighttime ozone. As the result, it is done with
 57 the use of the following system of equations:

$$58 \quad \begin{cases} P_{OH} = k_2 \cdot O(^3P) \cdot O_2 \cdot M - k_3 \cdot O(^3P) \cdot O_3 \\ VER_{2\mu m} = k_4(9,7) \cdot F_9(P_{OH}, M, T, O(^3P)) + k_4(8,6) \cdot F_8(P_{OH}, M, T, O(^3P)) \end{cases} \quad (4)$$

59 Thus, we derive the local values of $O(^3P)$, P_{OH} , and $OH(v=5-9)$ from SABER data with the use of eqs. (3-4) and apply sets of
 60 data (T , M , $OH(v=5-9)$, and $O(^3P)$) to retrieve the local concentrations of $O(^1D)$ with the use of eq. (2).

62 3 $O(^1D)$ nighttime distributions

63 We use the version 2.0 of the SABER data product (Level2A) for the simultaneously measured $VER_{2\mu m}$, O_3 , and T profiles
 64 within the 0.01–0.0001 hPa pressure (p) interval (approximately 80–105 km in 2003-2005). We take only nighttime data
 65 when the solar zenith angle $\chi > 95^\circ$. The range of latitudes covered by the satellite trajectory in a month was divided into 20
 66 bins $\sim (5.5-8)^\circ$ each. 1500-3000 single profiles of $O(^1D)$ concentration fall into one bin during a month of SABER
 67 observations. For each bin we calculate monthly averaged zonal mean $\langle O(^1D) \rangle$ distributions (hereafter, the angle brackets
 68 are used to denote timely and spatially averaged values). For annually averaged distributions, we use 40 bins $\sim 4^\circ$ each.

69 Monthly averaged $\langle O(^1D) \rangle$ distributions in corresponding month of 2003-2005 are shown in Figs. 1–3. Let’s analyze the
 70 presented data using the distributions in 2003 as an example. Depending on the range of latitudes covered by the satellite
 71 trajectory in specified month, the figures show from 2 to 4 maxima which are localized in height (at $\sim 92-96$ km) and latitude
 72 (at $\sim 20-40^\circ S, N$ and $\sim 60-80^\circ S, N$). The values of the maxima can reach up to 300 cm^{-3} and more in both hemispheres and
 73 different months, for example, in January-March and in May-August. Nevertheless, annual cycle of southern $O(^1D)$
 74 demonstrates certain differences from northern one, i.e. many features of $\langle O(^1D) \rangle$ in the southern hemisphere are not
 75 repeated in the northern hemisphere with a shift of 6 months. In particular, the distributions in January-February show 2
 76 pronounced maxima with close values (up to 300 cm^{-3}): the first one is at ~ 95 km and $\sim 50-60^\circ S$, the second one is at ~ 93 km
 77 and $\sim 60-80^\circ N$. Half year later (in July-August), we can see 1-2 weak maxima in the southern hemisphere and a strongly
 78 pronounced maximum at ~ 95 km and $\sim 40-50^\circ N$. The similar pattern can be noticed comparing the $\langle O(^1D) \rangle$ distributions



79 in June and December. The satellite trajectory in March and September allows us to observe simultaneously 4 maxima. Note
80 that the southern high-latitude maximum (up to 340 cm^{-3}) in March does not correspond to the relatively weak northern
81 high-latitude maximum in September.
82 The $\langle O(^1D) \rangle$ evolutions in 2004-2005 are very similar to 2003. Nevertheless, one can see some differences. First of all, in
83 January-February 2004, there is pronounced particularity above 60°N below 90 km which does not appear in 2003 and 2005.
84 Kulikov et al. (2019) found the similar features in the latitude dependence of nighttime ozone chemical equilibrium
85 boundary in January-March 2004 above 60°N and connected it with abnormal dynamics of stratospheric polar vortex during
86 2003-2004 Arctic winter. There are additional features also which take place in a specific year, but absent in other two
87 years. In particular, the northern high-latitude maximum in January-February 2003 is remarkably higher (by the value)
88 than the ones in January-February 2004-2005. The southern high-latitude maximum (up to 340 cm^{-3}) in March 2003
89 corresponds to the same maximum in March 2005 but both maxima are remarkably higher than the one in March 2004. The
90 reverse (relative to December 2003 and 2005) ratio can be observed for the values of southern and northern maxima in
91 December 2004.
92 Annually averaged $\langle O(^1D) \rangle$ distributions in 2003-2005 are shown in Fig. 4. There can be seen one weak maximum at
93 $\sim 93 \text{ km}$ and $\sim 65^\circ\text{S}$ with the values of $150\text{-}160 \text{ cm}^{-3}$ and 3 pronounced maxima (with the values up to 230 cm^{-3}) at $\sim 95 \text{ km}$
94 and $\sim 35^\circ\text{S}$, at $\sim 94 \text{ km}$ and $\sim 40^\circ\text{N}$, at $\sim 93 \text{ km}$ and $\sim 65\text{-}75^\circ\text{N}$ correspondingly. In general, there is slightly more $O(^1D)$ in the
95 northern hemisphere than in the southern hemisphere.
96 The systematic uncertainty of presented data is defined mainly by uncertainties in $VER_{2 \mu\text{m}}$, O_3 , T measurements, and in the
97 rates of chemical and physical processes included in the $\text{OH}(v)$ model. We repeated the analysis presented in Fytterer et al.
98 (2019) (see Sect. 3.4) and found that summarized uncertainty of $O(^1D)$ varied in the range of (37-52)% depending on the
99 pressure level. Due to averaging, the random error of presented data is negligible.

100

101 4 Discussion and Conclusion

102 According to various early papers (Nicolet, 1959; Ghosh & Gupta, 1970; Shimazaki & Laird, 1970; Harris & Adams, 1983),
103 daytime $O(^1D)$ concentrations at 90-100 km varied in the range of $(10^2\text{-}10^3) \text{ cm}^{-3}$. Brasseur & Solomon (2005) published the
104 table (see Table A.6.2.c) where daytime $O(^1D)$ changed from 70 cm^{-3} at 90 km to 140 cm^{-3} at 100 km. The presented results
105 show that monthly and annually mean nighttime $O(^1D)$ concentrations at these altitudes can reach 300 cm^{-3} and 200 cm^{-3} ,
106 respectively. Thus, nighttime concentrations of $O(^1D)$ are comparable with daytime concentrations of this component and, in
107 principle, can impact noticeably the chemistry and thermal balance of the mesopause region. More detailed analysis of this
108 impact should be carried out with the use of the global 3D chemical transport model of the mesosphere – lower
109 thermosphere.

110 **Data availability.** The SABER data used in this study can be downloaded from [ftp://saber.gats-](ftp://saber.gats-inc.com/Version2_0/Level2A/)
111 [inc.com/Version2_0/Level2A/](ftp://saber.gats-inc.com/Version2_0/Level2A/). The presented data can be downloaded from
112 http://www.iapras.ru/english/structure/dep_240/dep_240.html.

113 **Author contributions.** Both authors contributed equally to this paper.

114 **Competing interests.** The authors declare that they have no conflict of interest.

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116 References

- 117 Adler-Golden, S.: Kinetic parameters for OH nightglow modeling consistent with recent laboratory measurements, *J.*
118 *Geophys. Res.*, 102, 19969–19976, doi:10.1029/97JA01622, 1997.
- 119 Baasandorj, M., Hall, B. D., and Burkholder, J. B.: Rate coefficients for the reaction of O(¹D) with the atmospherically long-
120 lived greenhouse gases NF₃, SF₅CF₃, CHF₃, C₂F₆, c-C₄F₈, n-C₅F₁₂, and n-C₆F₁₄. *Atmos. Chem. Phys.*, 12, 11753–11764,
121 doi:10.5194/acp-12-11753-2012, 2012.
- 122 Brasseur, G. P., and Solomon, S.: *Aeronomy of the middle atmosphere: Chemistry and physics of the stratosphere and*
123 *mesosphere* (3rd ed). Dordrecht, Netherlands: Springer Science and Business Media, 2005.
- 124 Burkholder, J. B., Sander, S. P., Abbatt, J., Barker, J. R., Huie, R. E., Kolb, C. E., et al.: Chemical kinetics and
125 photochemical data for use in atmospheric studies, evaluation no. 18, JPL Publication 15–10, Pasadena, CA: Jet Propulsion
126 Laboratory, <http://jpldataeval.jpl.nasa.gov>, 2015.
- 127 Caridade, P. J. S. B., Horta, J.-Z. J., and Varandas, A. J. C.: Implications of the OCOH reaction in hydroxyl nightglow
128 modeling, *Atmos. Chem. Phys.*, 13, 1–13, doi:10.5194/acp-13-1-2013, 2013.
- 129 Fytterer, T., von Savigny, C., Mlynczak, M., and Sinnhuber, M.: Model results of OH airglow considering four different
130 wavelength regions to derive night-time atomic oxygen and atomic hydrogen in the mesopause region, *Atmos. Chem. Phys.*,
131 19, 1835–1851, doi:10.5194/acp-19-1835-2019, 2019.
- 132 Ghosh, S. N., and Gupta, S. K.: Altitude distributions of and radiations from certain oxygen and nitrogen metastable
133 constituents, *J. Geomagn. Geoelectr.*, 22, 329–339, 1970.
- 134 Harris, R. D., and Adams, G. W.: Where does the O(¹D) energy go? *J. Geophys. Res.*, 88(A6), 4918–4928,
135 doi:10.1029/JA088iA06p04918, 1983.
- 136 Hiraki, Y., Tong, L., Fukunishi, H., Nanbu, K., Kasai, Y., and Ichimura, A.: Generation of metastable oxygen atom O(¹D) in
137 sprite halos, *Geophys. Res. Lett.*, 31, L14105, doi:10.1029/2004GL020048, 2004.
- 138 Hofzumahaus, A., Lefer, B. L., Monks, P. S., Hall, S. R., Kylling, A., Mayer B. et al.: Photolysis frequency of O₃ to O(¹D):
139 Measurements and modeling during the International Photolysis Frequency Measurement and Modeling Intercomparison
140 (IPMMI), *J. Geophys. Res.*, 109, D08S90, doi:10.1029/2003JD004333, 2004.
- 141 Hunt, B. G.: A diffusive-photochemical study of the mesosphere and lower thermosphere and the associated conservation
142 mechanisms, *J. Atm. Terr. Phys.*, 33, 1869–1892, 1971.
- 143 Kalogerakis, K. S., Smith, G. P., and Copeland, R. A.: Collisional removal of OH(X ²Π, v = 9) by O, O₂, O₃, N₂, and CO₂. *J.*
144 *Geophys. Res.*, 116, D20307, doi:10.1029/2011JD015734, 2011.
- 145 Kalogerakis, K. S., Matsiev, D., Sharma, R. D., and Wintersteiner, P. P.: Resolving the mesospheric nighttime 4.3 μm
146 emission puzzle: Laboratory demonstration of new mechanism for OH(*v*) relaxation, *Geophys. Res. Lett.*, 43, 8835–8843,
147 doi:10.1002/2016GL069645, 2016.
- 148 Kalogerakis, K. S.: A previously unrecognized source of the O₂ atmospheric band emission in earth's nightglow. *Science*
149 *Advances*, 5, eaau9255, doi:10.1126/sciadv.aau9255, 2019.
- 150 Kulikov, M. Y., Belikovich, M. V., Grygalashvyly, M., Sonnemann, G. R., Ermakova, T. S., Nechaev, A. A., and Feigin, A.
151 M.: Nighttime ozone chemical equilibrium in the mesopause region, *J. Geophys. Res.*, 123, 3228–3242,
152 doi:10.1002/2017JD026717, 2018.
- 153 Kulikov, M. Yu., Nechaev, A. A., Belikovich, M. V., Vorobeva, E. V., Grygalashvyly, M., Sonnemann, G. R., and Feigin,
154 A. M.: Border of nighttime ozone chemical equilibrium in the mesopause region from saber data: implications for derivation
155 of atomic oxygen and atomic hydrogen, *Geophys. Res. Lett.*, 46, 997–1004, doi:10.1029/2018GL080364, 2019.
- 156 Mlynczak, M. G., Hunt, L. A., Mast, J. C., Marshall, B. T., Russell III, J. M., Smith, A. K., Siskind, D. E., Yee, J.-H.,
157 Mertens, C. J., Martin-Torres, F. J., Thompson, R. E., Drob, D. P., and Gordley, L. L.: Atomic oxygen in the mesosphere and



- 158 lower thermosphere derived from SABER: Algorithm theoretical basis and measurement uncertainty, *J. Geophys. Res.*, 118,
159 5724–5735, doi:10.1002/jgrd.50401, 2013.
- 160 Mlynczak, M. G., Hunt, L. A., Russell, J. M., III, and Marshall, B. T.: Updated SABER night atomic oxygen and
161 implications for SABER ozone and atomic hydrogen, *Geophys. Res. Lett.*, 45, 5735–5741, doi:10.1029/2018GL077377,
162 2018.
- 163 Nicolet, M.: The constitution and composition of the upper atmosphere, *Proc. IRE*, 47, 142–147, 1959.
- 164 Panka, P. A., Kutepov, A. A., Kalogerakis, K. S., Janches, D., Russell, J. M., Rezac, L., Feofilov, A. G., Mlynczak, M. G.,
165 and Yigit, E.: Resolving the mesospheric nighttime 4.3 μm emission puzzle: Comparison of the $\text{CO}_2(\nu_3)$ and $\text{OH}(\nu)$
166 emission models, *Atm. Chem. Phys.*, 17, 9751–9760, doi:10.5194/acp-17-9751-2017, 2017.
- 167 Sharma, R. D., Wintersteiner, P. P., and Kalogerakis, K. S.: A new mechanism for OH vibrational relaxation leading to
168 enhanced CO_2 emissions in the nocturnal mesosphere, *Geophys. Res. Lett.*, 42, 4639–4647, doi:10.1002/2015GL063724,
169 2015.
- 170 Shepherd, M., Shepherd, G., and Codrescu, M.: Perturbations of $\text{O}(^1\text{D})$ VER, temperature, winds, atomic oxygen, and TEC
171 at high southern latitudes, *J. Geophys. Res.*, 124, 4773–4795, doi:10.1029/2019JA026480, 2019.
- 172 Shimazaki, T., and Laird, A. R.: A model calculation of the diurnal variation in minor neutral constituents in the mesosphere
173 and lower thermosphere including transport effects, *J. Geophys. Res.*, 75, 3221–3235, doi:10.1029/JA075i016p03221, 1970.
- 174 Taniguchi, N., Hayashida, S., Takahashi, K., and Matsumi, Y.: Sensitivity studies of the recent new data on $\text{O}(^1\text{D})$ quantum
175 yields in O_3 Hartley band photolysis in the stratosphere, *Atmos. Chem. Phys.*, 3, 1293–1300, doi:10.5194/acp-3-1293-2003,
176 2003.
- 177 Varandas, A. J. C.: Reactive and non-reactive vibrational quenching in OCOH collisions, *Chem. Phys. Lett.*, 396, 182–190,
178 doi:10.1016/j.cplett.2004.08.023, 2004.
- 179 Xu, J., Gao, H., Smith, A. K., and Zhu, Y.: Using TIMED/SABER nightglow observations to investigate hydroxyl emission
180 mechanisms in the mesopause region, *J. Geophys. Res.*, 117, D02301, doi:10.1029/2011JD016342, 2012.
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183 **Table 1. List of processes.**

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	Process	Rate	Reference
1	$\text{H} + \text{O}_3 \rightarrow \text{O}_2 + \text{OH}(v)$	$k_1, k_1(v)$	Burkholder et al. (2015), Alder-Golden (1997, Table 1)
2	$\text{O}(^3\text{P}) + \text{O}_2 + \text{M} \rightarrow \text{O}_3 + \text{M}$	k_2	Burkholder et al. (2015)
3	$\text{O}(^3\text{P}) + \text{O}_3 \rightarrow 2\text{O}_2$	k_3	Burkholder et al. (2015)
4	$\text{OH}(v) \rightarrow \text{OH}(v')$	$k_4(v, v')$	Xu et al. (2012, Table A1)
5	$\text{OH}(v) + \text{N}_2 \rightarrow \text{OH}(v') + \text{N}_2$	$k_5(v, v')$	Adler-Golden (1997, Table 1), Kalogerakis et al. (2011)
6	$\text{OH}(v) + \text{O}_2 \rightarrow \text{OH}(v') + \text{O}_2$	$k_6(v, v')$	Adler-Golden (1997, Table 3), corrected and adjusted by Fyterer et al. (2019)
7	$\text{OH}(v) + \text{O}(^3\text{P}) \rightarrow \text{H} + \text{O}_2$	$k_7(v)$	Varandas (2004, Table 3, M I)
8	$\text{OH}(v) + \text{O}(^3\text{P}) \rightarrow \text{OH}(v') + \text{O}$	$k_8(v, v')$	Caridade et al. (2013, Table 1)
9	$\text{OH}(9) + \text{O}(^3\text{P}) \rightarrow \text{OH}(3,4) + \text{O}(^1\text{D})$	k_9	Fyterer et al. (2019)
10	$\text{OH}(8) + \text{O}(^3\text{P}) \rightarrow \text{OH}(3) + \text{O}(^1\text{D})$	k_{10}	Fyterer et al. (2019)
11	$\text{OH}(7) + \text{O}(^3\text{P}) \rightarrow \text{OH}(\leq 2) + \text{O}(^1\text{D})$	k_{11}	Fyterer et al. (2019)
12	$\text{OH}(6) + \text{O}(^3\text{P}) \rightarrow \text{OH}(\leq 1) + \text{O}(^1\text{D})$	k_{12}	Fyterer et al. (2019)
13	$\text{OH}(5) + \text{O}(^3\text{P}) \rightarrow \text{OH} + \text{O}(^1\text{D})$	k_{13}	Fyterer et al. (2019)
14	radiative decay of $\text{O}(^1\text{D})$	k_{14}	Burkholder et al. (2015)
15	$\text{O}(^1\text{D}) + \text{O}_2 \rightarrow \text{O}(^3\text{P}) + \text{O}_2$	k_{15}	Burkholder et al. (2015)
16	$\text{O}(^1\text{D}) + \text{N}_2 \rightarrow \text{O} + \text{N}_2$	k_{16}	Burkholder et al. (2015)

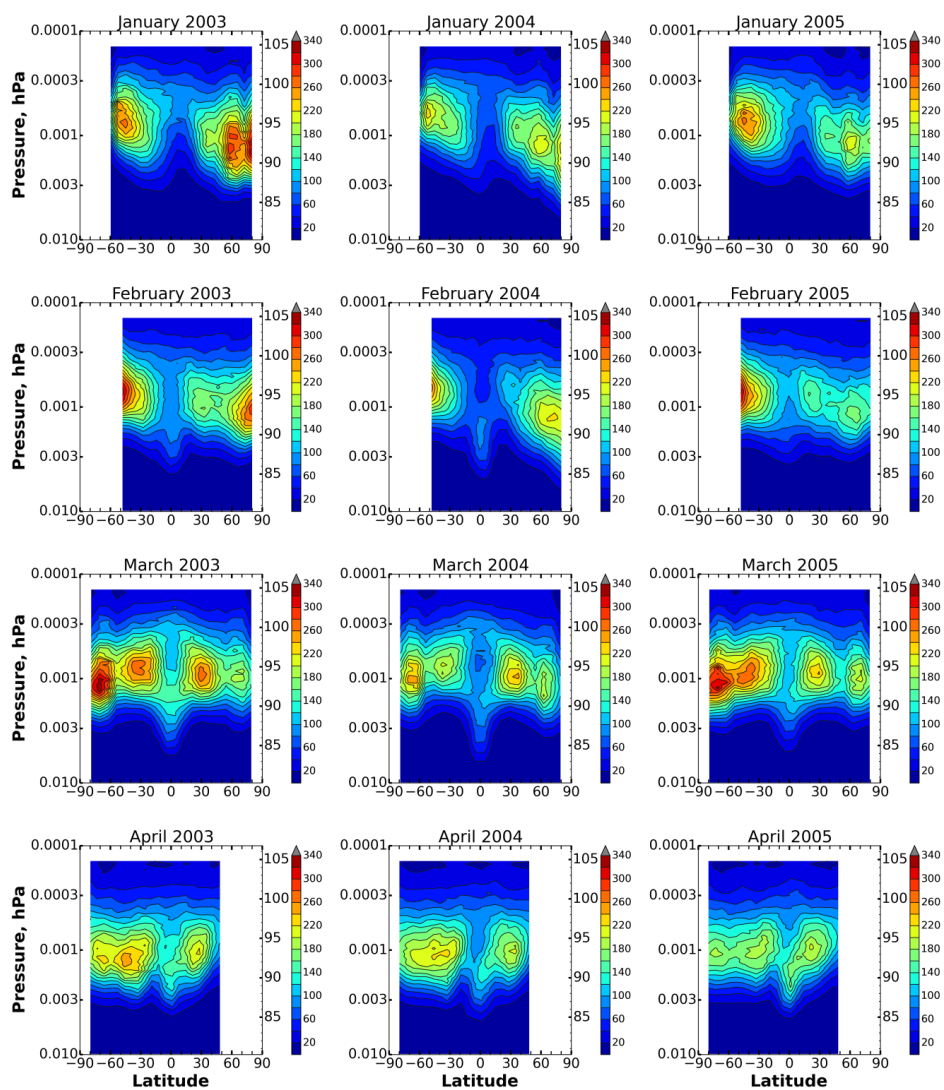
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187 **Figures**

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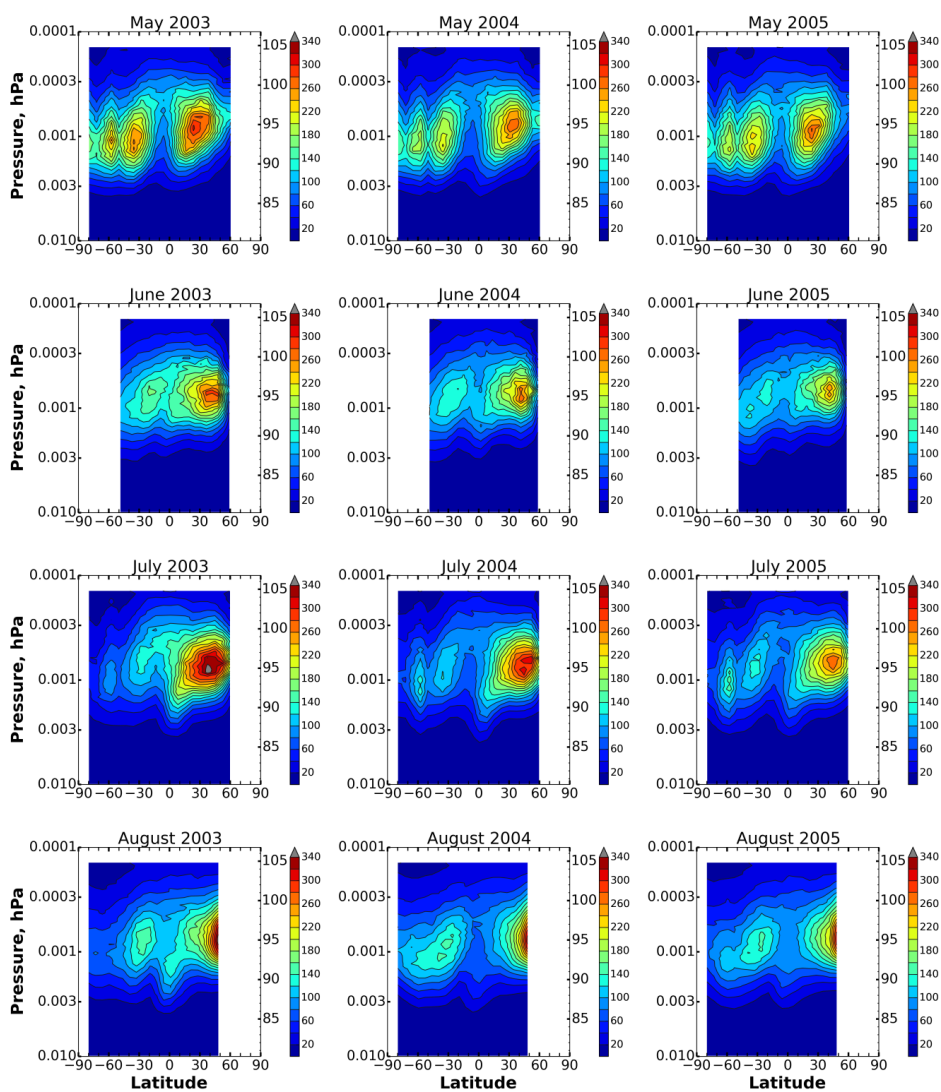
Figure 1. Monthly averaged $O(^1D)$ concentration (in cm^{-3}) in January-April of 2003-2005.

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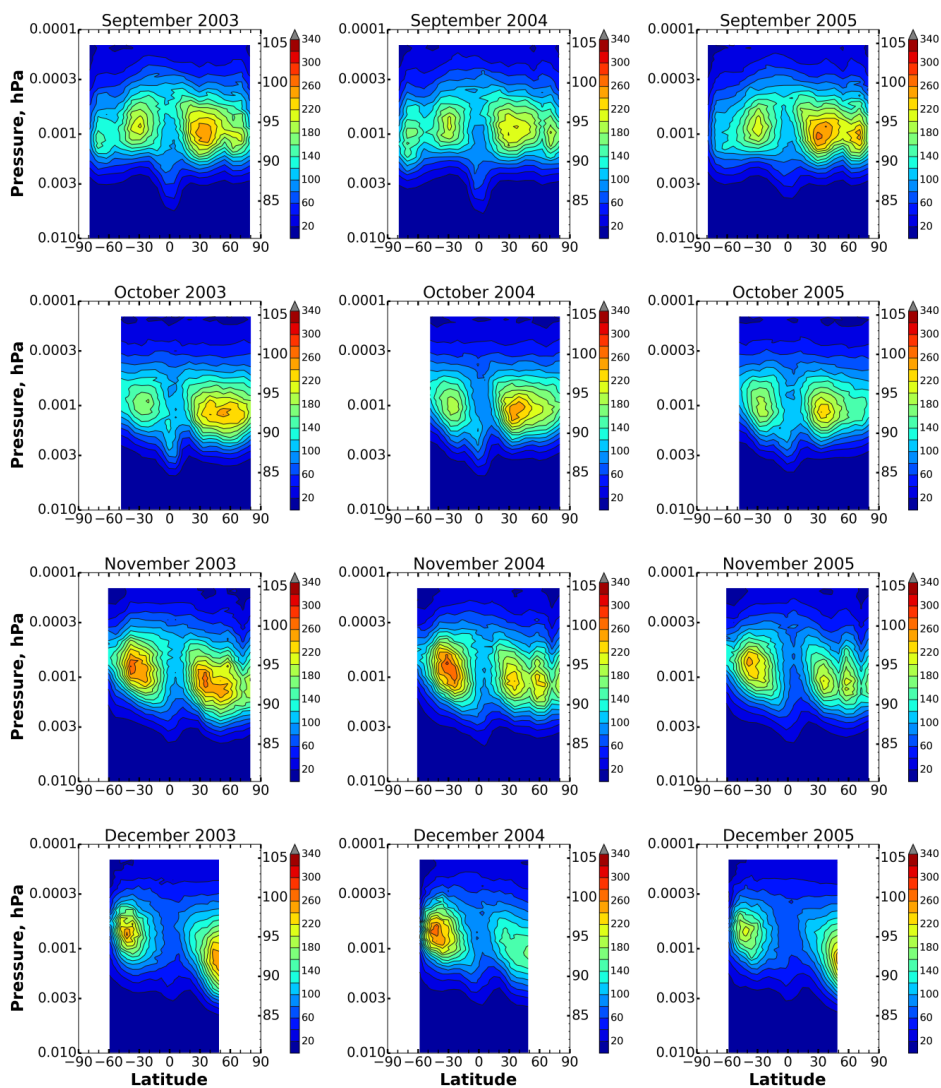
196 **Figure 2. Monthly averaged O(¹D) concentration (in cm⁻³) in May-August of 2003-2005.**

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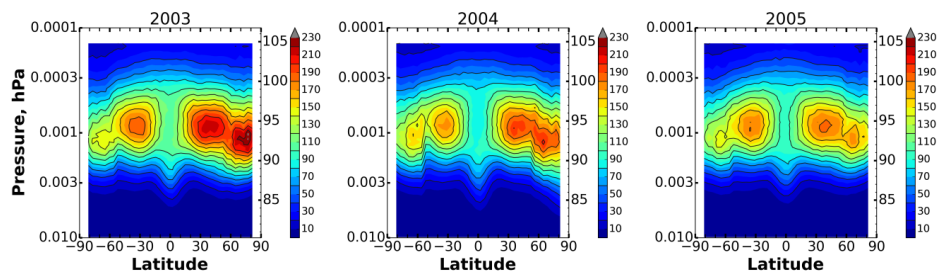
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201 **Figure 3. Monthly averaged O(¹D) concentration (in cm⁻³) in September-December of 2003-2005.**

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205 **Figure 4. Annually averaged O(¹D) concentration (in cm⁻³) in 2003-2005.**