

New electron energy transfer and cooling rates by excitation of O₂

A. V. Pavlov

Institute of Terrestrial Magnetism, Ionosphere and Radio-Wave Propagation, Russia Academy of Science (IZMIRAN), Troitsk, Moscow Region, 142092, Russia

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Abstract. In this work I present the results of a study of the electron cooling rate, the production rates of vibrationally excited O₂, and the production frequency of the O₂ vibrational quanta arising from the collisions of electrons with O₂ molecules as functions of the electron temperature. The electron energy transfer and cooling rates by vibrational excitation of O₂ have been calculated and fitted to analytical expressions by use of the revised vibrationally excited O₂ cross sections. These new analytical expressions are available to the researcher for quick reference and accurate computer modeling with a minimum of calculations. It is also shown that the currently accepted rate of electron energy loss associated with rotational transitions in O₂ must be decreased by a factor of 13.

solar-activity conditions the increase in the O⁺(⁴S) + O₂ loss rate due to vibrationally excited O₂ produces up to 40% reductions in the calculated daytime F2 peak electron density if the vibrational temperature of O₂ is the same as the neutral translation temperature. To calculate real values of the O₂ vibrational temperatures and the effects of vibrationally excited O₂ on the calculated electron density the knowledge of the energy transfer rates from thermal electrons to vibrationally excited O₂ is required.

The energy transfer rates from electrons to O₂ have been calculated and fitted to analytical expressions by Prasad and Furman (1973) for vibrational and low-lying electronic excitation of O₂. These electron cooling rates due to vibrational excitation of O₂ and excitation of ¹Δ_g and ¹Σ_g⁺ states of O₂ which are currently used in models of the ionosphere are based on the pioneering O₂ vibrational and electronic cross sections of Linder and Schmidt (1971). However, Lawton and Phelps (1978) found that they have to increase the O₂ energy integrated vibrational cross sections of Linder and Schmidt (1971) by a factor of 2 to obtain satisfactory agreement with their measured electron transport coefficients. The recent measurements of the energy integrated cross sections of Allan (1995) are larger than those given by Linder and Schmidt (1971) by a factor of nearly 3. Linder and Schmidt (1971) measured the energy integrated cross sections only for excitation of O₂ at the vibrational levels $v = 1, 2, 3, 4$ while Allan (1995) presented these cross sections for vibrational excitation of the $v = 0 \rightarrow 1, 2, 3, 4, 5, 6, 7$ levels by electrons with energies in the range from threshold to 3 eV where the ²Π_g resonance dominates. Furthermore, as a result of the improved sensitivity and resolution of the instrument, the number of the narrow resonance peaks given by Allan (1995) for each vibrational level $v = 1, 2, 3, 4$ is larger than those obtained by Linder and Schmidt (1971). The primary object of this study is to use the O₂ energy integrated vibrational cross sections measured by Allan (1995) to calculate and to fit to new analytical expressions for energy transfer rates from electrons to O₂.

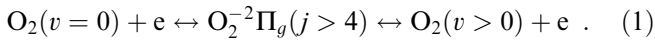
1 Introduction

The electron temperature in the ionosphere is of great significance in that it usually controls the rates of many physical and chemical ionospheric processes. The theoretical computation of electron temperature distribution in the ionosphere requires the knowledge of various heating and cooling rates. Lane and Dalgarno (1969) have shown that the electron cooling by excitation of O₂ should be important in the E-region of the ionosphere (we shall use the symbol O₂ for the ground electronic state X³Σ_g⁻ of O₂). Hierl *et al.* (1997) found that the big difference between the high temperature flowing afterglow and drift tube measurements (McFarland *et al.*, 1973; Albritton *et al.*, 1977) of the effective rate coefficient for the reaction of O⁺(⁴S) with O₂ results from the input of the reactions between the vibrationally excited O₂ and O⁺(⁴S). Pavlov *et al.* (1998) and Pavlov (1998) found that at solar maximum and moderate

As far as the author knows, rotational excitation cross sections of O₂ have not been measured. To calculate the electron cooling rate due to rotational excitation of O₂ by electron impact the quadrupole Born approximation formula of Mentzoni and Row (1963) with theoretical rotational excitation cross sections given by Gerjuoy and Stein (1955) is usually used. As a result, the electron cooling rate given by Dalgarno *et al.* (1968) with an effective quadrupole moment of Hake and Phelps (1967) is currently used in models of the ionosphere. The accuracy of this approach is discussed in our work on the basis of the Smith and Dean (1975) measurements of the electron cooling rate in O₂ gas.

2 Electron energy transfer rates for vibrational excitation of O₂

During the vibrational excitation of the unexcited O₂($v=0$) by electron impact the temporary capture of the electron occurs to form the excited ion O₂⁻²Π_g at vibrational level, j , with the following vibrational excitation of O₂ :



The electron cooling rate arising from the collisions of thermal electrons with O₂($v=0$) can be calculated as

$$L(v=0) = N_e \sum_{v=1}^7 Q_{0v} \{ [\text{O}_2(0)] - [\text{O}_2(v)] \exp(E_v T_e^{-1}) \} , \quad (2)$$

where

$$Q_{0v} = E_v \{ 8kT_e (\pi m_e)^{-1} \}^{0.5} \int_0^{\infty} \sigma_{0v}(x) x \exp(-x) dx , \quad (3)$$

$E_v = vE_1 - v(v-1)\Delta E$, $x = E(kT_e)^{-1}$, E is the energy of electrons, E_v is the energy of the v -th vibrational level, $\Delta E = 17.3\text{K}$ is anharmonicity of O₂ (Radzig and Smirnov, 1980), $E_1 = 2239\text{K}$ (or 0.1929 eV) is the energy of the first vibrational level of O₂ given by Radzig and Smirnov (1980), $\sigma_{0v}(E)$ are the partial cross sections for excitation of O₂(0) by electrons, k is Boltzmann's coefficient, and m_e denotes the mass of electrons.

Allan (1995) found very sharp peaks in the vibrational cross sections in the resonance region 0.3–3.0 eV of O₂ and could determine the cross sections integrated over a certain range of energy for vibrational excitation of O₂. The use of the integral vibrational-excitation cross sections of Allan (1995) in these narrow spikes leads to

$$Q_{0v} = E_v \sqrt{8} (kT_e \pi m_e)^{-0.5} \sum_{j=5}^{24} S_{jv} X_j \exp(-X_j) , \quad (4)$$

where j is the number of the O₂⁻²Π_g vibrational level, S_{jv} is the integral vibrational-excitation cross section for the sharp peak in the vibrational cross section at the energy E_j^* , $X_j = E_j^*(kT_e)^{-1}$, E_j^* is the energy of the incident electron at the centre of the O₂⁻²Π_g resonance given

by Allan (1995), the energy resolution is not sufficient to resolve rotational structure and the discussed measurements of cross sections yield rotationally averaged vibrational-excitation cross sections.

Equation (2) includes both excitation and deexcitation, which are related through the principle of detailed balancing. To calculate [O₂(v)] it is necessary to solve the vibrational quanta continuity equation [a Boltzmann distribution of O₂(v)] or the full continuity equations for densities of O₂(v) [a non-Boltzmann distribution of O₂(v)]. The system of full continuity equations and the vibrational quanta continuity equation are described by Pavlov (1989, 1994, 1998) using the harmonic and anharmonic oscillator energy level approximations, the vibrational-vibrational and vibrational-translational energy exchange of O₂(v), the diffusion of O₂(v) in the mixture of O₂(0), N₂, and O, and the production rates of O₂(v). The values of the production rates of O₂(v) due to the thermal electron excitation of O₂(0) and the deexcitation of O₂(v) in the reaction (1) can be calculated as

$$q_v = N_e Q_{0v} E_v^{-1} \{ [\text{O}_2(0)] - [\text{O}_2(v)] \exp(E_v T_e^{-1}) \} \quad (5)$$

where Q_{0v} is the same as in Eq. (4).

Pavlov (1998) found that during the undisturbed and geomagnetic storm periods at middle latitudes at moderate solar-activity conditions deviation from a Boltzmann distribution of O₂(v) was large at the vibrational energy levels $v > 3$ and close to a Boltzmann distribution of O₂(v) for $v = 1, 2$, and 3. The main contribution to $L(v=0)$ arising from the deexcitation of O₂(v) is brought by the deexcitation of O₂(1) and O₂(2). As a result, it is possible to use a Boltzmann distribution of O₂(v) in calculations of $L(v=0)$ as

$$\begin{aligned} [\text{O}_2(v)] &= [\text{O}_2(0)] \{ \alpha(1 + \alpha)^{-1} \}^v \\ &= [\text{O}_2(0)] \exp(-vE_1/T_{\text{vib}}) , \end{aligned} \quad (6)$$

where the vibrational quanta is defined as $\alpha = \sum_{v=1}^{\infty} v [\text{O}_2(v)] / [\text{O}_2]$, and the vibrational temperature is defined as $T_{\text{vib}} = -E_1 / \ln[\alpha(1 + \alpha)^{-1}]$.

In this case from Eq. (6) it follows that the sum of the vibrationally unexcited and excited molecular oxygen concentrations is

$$[\text{O}_2] = \sum_{v=0}^{\infty} [\text{O}_2(v)] = [\text{O}_2(0)] (1 + \alpha) . \quad (7)$$

There is no possibility to differentiate the vibrationally unexcited O₂ from the vibrationally excited O₂ during the mass-spectrometer measurements of the neutral species in the atmosphere, and it is necessary to point out that the MSIS-86 model (Hedin, 1987) which is usually used in most models calculates the value of [O₂]. The difference between [O₂] and [O₂(0)] is not negligible during solar maximum. For example, [O₂]/[O₂(0)] = 1.29 for $T_{\text{vib}} = 1500\text{K}$.

The total electron cooling rate associated with vibrational transitions of O₂ arises from the collisions of thermal electrons with unexcited O₂($v=0$) and vibrationally excited O₂($v > 0$) molecules. The value of this rate is defined as $L_{\text{vib}}(\text{O}_2) = \sum_{v=0}^{\infty} L(v)$. As far as the

author knows, vibrational excitation cross sections of O₂($v > 0$) by electron impact have not been measured and not calculated from theory. To simplify this problem Prasad and Furman (1973) did not distinguish between vibrational excitation cross sections of O₂($v = 0$) and O₂($v > 0$). In this approximation for the Boltzmann distribution given by Eq. (6) the total rate of electron energy loss associated with vibrational transitions of O₂(v) is

$$L_{\text{vib}}(\text{O}_2) = N_e[\text{O}_2] \sum_{v=1}^7 Q_{0v} \{1 - \exp[vE_1(T_e^{-1} - T_{\text{vib}}^{-1})]\} . \quad (8)$$

The number density of vibrationally unexcited O₂ is much larger than the number density of vibrationally excited O₂ during the undisturbed and geomagnetic storm periods at middle latitudes (Pavlov, 1989, 1994, 1998). As a result, the collisions of thermal electrons with vibrationally unexcited O₂ yield the main contribution to $L_{\text{vib}}(\text{O}_2)$ and the possible error of Eq. (8) is expected to be small.

To calculate α or T_{vib} it is necessary to solve the vibrational quanta continuity equation (Pavlov, 1989, 1994, 1998). The vibrational-translational energy exchange of O₂(v), the diffusion of O₂(v) in the mixture of O₂(0), N₂, and O, and the production frequencies of α are taken into account in the vibrational quanta continuity equation. The values of the production frequency, \bar{W} , of the vibrational quanta due to the thermal electron excitation of O₂(0) and the deexcitation of N₂(v) can be calculated as $\bar{W} = L(v=0)\{E_1[\text{O}_2]\}^{-1}$.

Prasad and Furman (1973) found the analytical expressions for Q_{0v} (PF) based on the O₂ integral vibrational cross sections of Linder and Schmidt (1971) for the vibrational levels $v = 1, 2$ and 3 and we can compare their results with Q_{0v} based on the Allan (1995) cross sections.

Solid lines on Fig. 1 show Q_{0v} as functions of T_e and v . Curves 1, 2, ... 7 on Fig. 1 are the calculated Q_{0v} for $v = 1, 2, \dots, 7$ of O₂(v). Dashed lines 1, 2 and 3 on Fig. 1 show Q_{0v} (PF) for $v = 1, 2$ and 3 of O₂(v). It is evident that the energy transfer rates obtained by using the Allan (1995) cross sections are much larger than those obtained by using the Linder and Schmidt (1971) data within the electron temperature range 300–6000 K: Q_{01}/Q_{01} (PF) = 4.9–12.4, Q_{02}/Q_{02} (PF) = 4.0–10.6, and Q_{03}/Q_{04} (PF) = 4.6 – 69.1. These results show the importance of the use of the integral vibrational cross sections of Allan (1995) instead of those from Linder and Schmidt (1971) in calculations of the production rates of O₂(v) due to the thermal electron excitation of O₂(0).

The total cooling rate by vibrational excitation of O₂ is given by Prasad and Furman (1973) as

$$L(\text{PF}) = N_e[\text{O}_2]U(T_e)\{1 - \exp[2770(T_e^{-1} - T_n^{-1})]\}, \quad (9)$$

where $U(T_e) = 5.196 \cdot 10^{-13} \exp[f_4(1/700 - 1/T_e)]$, $f_4 = 3300 - 839 \sin[0.000191(T_e - 2700)]$, T_n is a neutral temperature, the units of T_e and T_n are K and the units of $U(T_e)$ are eV cm³s⁻¹. The value of $L(\text{PF})$ includes both excitation ($U(T_e)$) and deexcitation ($U(T_e)\exp[2770(T_e^{-1} - T_n^{-1})]$) of O₂(v). It can be seen from the comparison of Eqs. (8) and (9) that the analytical expression for the cooling rate was found by Prasad and Furman (1973) by assuming that the value of the deexcitation rate of O₂($v > 1$) is the same as that for O₂($v = 1$) and $T_{\text{vib}} = T_n$. This assumption leads to the most simple approach in calculations of the cooling rate given by Eq. (8) as

$$L_{\text{vib}}(\text{O}_2) = N_e[\text{O}_2]Q(T_e)\{1 - \exp[E_1(T_e^{-1} - T_n^{-1})]\} , \quad (10)$$

where $Q(T_e) = \sum_{v=1}^7 Q_{0v}(T_e)$.

To compare $L_{\text{vib}}(\text{O}_2)$ with $L(\text{PF})$ it is necessary to compare $U(T_e)$ with $Q(T_e)$. The calculated values of

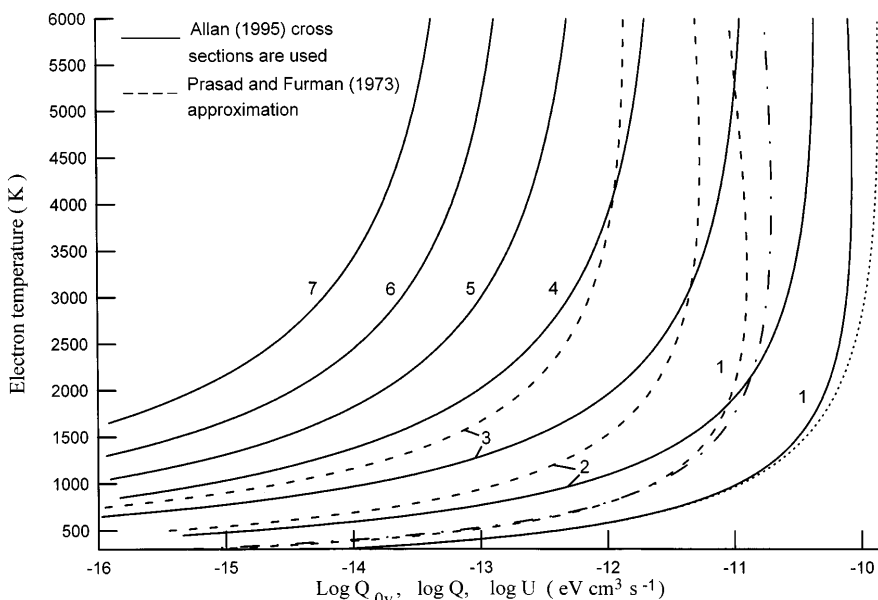


Fig. 1. Comparison of Q_{0v} calculated by use of the energy integrated cross sections of Allan (1995) (solid lines) and Linder and Schmidt (1971) (dashed lines) for the O₂ vibrational levels $v = 1, 2, \dots, 7$ (curves 1, 2, ... 7). Dotted line is $Q(T_e) = \sum_{v=1}^7 Q_{0v}(T_e)$ calculated in this study by using the energy integrated cross sections of Allan (1995). Dashed-dotted line is $U(T_e) = \sum_{v=1}^4 Q_{0v}(T_e)$ given by Prasad and Furman (1973) and based on the O₂ integral vibrational cross sections of Linder and Schmidt (1971)

$U(T_e)$ and $Q(T_e)$ are shown in Fig. 1 as dashed-dotted and dotted lines. The value of the ratio $U(T_e)/Q(T_e)$ is about 4.7–17.9 within the electron temperature range 300–6000 K. We can see that the use of the Prasad and Furman (1973) approach leads to very serious error in the cooling rate by vibrational excitation of O₂.

The value of Q_{0v} increases with increasing the number of vibrational level. The relative contribution to L and W of the upper vibrational levels can be evaluated from the ratio $R_k = \sum_{v=k}^7 Q_{0v}(T_e)/Q(T_e)$. The value of R_5 is less than 0.005 and the value of R_4 is less than 0.02 within the electron temperature range 300–6000 K. Therefore the vibrational levels $v = 1 - 4$ are enough to calculate L, W and T_{vib} . The values of $Q_{0v}(v > 4)$ are necessary to calculate [O₂($v > 4$)] by solving the full continuity equations for densities of O₂(v) given by Pavlov (1998) (a non-Boltzmann distribution of O₂(v)).

Wong *et al.* (1973), Shyn and Sweeney (1993), Allan (1995) and Noble *et al.* (1996) found the broad maximum in the vibrational cross sections of O₂ near 10 eV in addition to the sharp peaks below about 2.5 eV arising from the $^2\Pi_g$ resonance of the reaction (1). The $^2\Pi_g$ resonance dominates the vibrational cross sections in the ground electronic state of O₂ at low energies below about 2.5 eV while the $^4\Sigma_u^-$ and $^2\Sigma_u^-$ resonances of O₂ give the major contribution to the O₂ vibrational cross sections above 3–4 eV (Allan, 1995; Noble *et al.*, 1996).

Let us denote the value of Q_{0v} for the $^2\Pi_g$ resonance as $Q_{0v}(\Pi)$ ($Q_{0v}(\Pi)$ is presented in Fig. 1 as a function of T_e for the each vibrational level $v = 1 - 7$) and the value of Q_{0v} for the $^4\Sigma_u^-$ and $^2\Sigma_u^-$ resonances as $Q_{0v}(\Sigma)$. To calculate $Q_{0v}(\Sigma)$ from Eq. (3) we used the measured $\sigma_{0v}(E)$ of Noble *et al.* (1996) for the electron energy range 5–15 eV. We also used the differential cross sections, Ω_{0v} , given by Allan (1995) and $\sigma_{0v}(E^* = 5 - 7\text{eV})$ measured by Noble *et al.* (1996) to calculate $\sigma_{0v}(E) = \sigma_{0v}(E^*)\Omega_{0v}(E)/\Omega_{0v}(E^*)$ for the region of electron energy from 3 eV to 5 eV. As a result, the ratio $Q_{0v}(\Sigma)/Q_{0v}(\Pi)$ is less than 0.005 for $v = 1 - 4$ within the electron temperature range 300–5000 K and the effects of the $^4\Sigma_u^-$ and $^2\Sigma_u^-$ resonances on Q_{0v} are negligible in comparison with those for the $^2\Pi_g$ resonance.

The electron cooling rate, the production rates of O₂(v), and the production frequency of the O₂ vibrational quanta depend on Q_{0v} which is a function of the electron temperature. We now give an analytical expression, Q_{0v}^* , for Q_{0v} available to the researcher for

quick reference and accurate computer modeling with a minimum of calculations as

$$Q_{0v}^* = A_v \exp\{(1 - B_v T_e^{-1})(C_v + D_v \sin[F_v(T_e - G_v)])\} , \quad (11)$$

where the constants A_v, B_v, C_v, D_v, F_v , and G_v are given in Table 1.

To establish how good or bad our analytical expression can be we define quantitative measure of maximum error of analytical approach of Eq. (11) for given range of T_e as $\delta_v = \max |1 - Q_{0v}^*/Q_{0v}|$. The calculated values of δ_v are presented in Table 1 for the electron temperature range from 300–340 K to 3950–5650 K where Eq. (11) is recommended for use. The above analytical expressions reproduce the numerically calculated Q_{0v} within 9–14%. The measured energy integrated cross sections are believed to be accurate within $\pm 35\%$, except small amount of data where the error can be up to a factor of 2 (Allan, 1995). Therefore the accuracy within 9–14% given in Table 1 is enough to reproduce Q_{0v} .

To use Eq. (10) in calculations of the cooling rate by vibrational excitation of O₂ we obtained an analytical expression, $Q^*(T_e)$, for $Q(T_e)$ as

$$Q^*(T_e) = 8.8 \cdot 10^{-15} \exp\{(1 - 304 T_e^{-1}) \times (10.2 + 0.4 \sin[1.083 \cdot 10^{-3}(T_e - 1000)])\} , \quad (12)$$

eVcm³s⁻¹ .

The value of maximum error for the electron cooling rate may be evaluated as $\delta = \max |1 - Q^*(T_e)/Q(T_e)|$. We found that $\delta = 0.10$ within the electron temperature range 300–3600 K and this accuracy is enough to reproduce $L_{\text{vib}}(\text{O}_2)$.

3 The electron cooling rate by low-lying electronic excitation of O₂

There are several independent measurements of the integral excitation cross sections for the lower lying excited states of O₂($a^1\Delta_g$) and O₂($b^1\Sigma_g^+$) in the electron energy range up to 4–20 eV. Linder and Schmidt (1971) reported the integral cross sections for electronic excitation of the $a^1\Delta_g$ and $b^1\Sigma_g^+$ states with electrons of incident energies in the range from thresholds to 4 eV while Trajmar *et al.* (1971) measured these cross sections in the energy range 4–45 eV. The most recent measurements of the $a^1\Delta_g$ and $b^1\Sigma_g^+$ integral cross sections were

Table 1. Coefficients for calculations of Q_{0v}

v	A_v , eVcm ³ s ⁻¹	B_v , K	C_v	D_v	F_v	G_v , K	δ_v
1.	$8.56 \cdot 10^{-15}$	303	10.00	-0.20	$1.05 \cdot 10^{-3}$	3150	0.11 (300 K $\leq T_e \leq$ 5400 K)
2.	$1.15 \cdot 10^{-18}$	305	18.78	-0.25	$9.24 \cdot 10^{-4}$	3450	0.11 (340 K $\leq T_e \leq$ 5650 K)
3.	$1.77 \cdot 10^{-23}$	301	29.00	-0.20	$6.16 \cdot 10^{-4}$	3150	0.11 (300 K $\leq T_e \leq$ 5650 K)
4.	$7.05 \cdot 10^{-25}$	301	30.07	0.31	$1.00 \cdot 10^{-3}$	1800	0.11 (300 K $\leq T_e \leq$ 4550 K)
5.	$2.14 \cdot 10^{-28}$	299	37.52	-0.42	$5.28 \cdot 10^{-4}$	3000	0.11 (300 K $\leq T_e \leq$ 4000 K)
6.	$2.94 \cdot 10^{-31}$	300	43.27	-0.39	$7.85 \cdot 10^{-4}$	3000	0.09 (300 K $\leq T_e \leq$ 5600 K)
7.	$1.08 \cdot 10^{-35}$	299	53.10	-0.90	$5.70 \cdot 10^{-4}$	2900	0.14 (315 K $\leq T_e \leq$ 3950 K)

made by Middleton *et al.* (1992) in the electron energy range 4–45 eV, and Shyn and Sweeney (1993) for electron impact energies 5, 7, 10, and 15 eV. These measurements are in good agreement with each other at low energies below about 5–6 eV. The earlier measurements of Trajmar *et al.* (1971) are in disagreement with the data of Middleton *et al.* (1992) and Shyn and Sweeney (1993) close to and within the resonance peak region in the cross section of the $a^1\Delta_g$ and $b^1\Sigma_g^+$ states.

The cooling rates due to the excitation of the $a^1\Delta_g$ and $b^1\Sigma_g^+$ electronic states of O₂ by electron impact have been calculated and fitted to analytical expressions by Prasad and Furman (1973) by using the integral cross sections of Linder and Schmidt (1971). We found that the effects of the excitation of the $a^1\Delta_g$ and $b^1\Sigma_g^+$ electronic states of O₂ with electrons of incident energies in the range from thresholds to 4 eV give the main contribution to the electron cooling rates. As a result, we can conclude that the generally accepted electron cooling rates of Prasad and Furman (1973) due to the excitation of O₂($a^1\Delta_g$) and O₂($b^1\Sigma_g^+$) can be used in models of the ionosphere.

The electron cooling rates of Prasad and Furman (1973) includes both excitation and deexcitation of O₂($a^1\Delta_g$) and O₂($b^1\Sigma_g^+$), which are related through the principle of detailed balancing. To take into account deexcitation of O₂($a^1\Delta_g$) and O₂($b^1\Sigma_g^+$) Prasad and Furman (1973) assumed that $[O_2(a^1\Delta_g)] = [O_2(b^1\Sigma_g^+)] = [O_2] \exp(-11400 T_n^{-1})$ and $g_1/g_0 = g_2/g_0 = 1$ where g_0, g_1 , and g_2 are statistical weights of the electron levels $X^3\Sigma_g^-, a^1\Delta_g$, and $b^1\Sigma_g^+$ of O₂. After correction of these deexcitation terms the electron cooling rates of Prasad and Furman (1973) are given as

$$L(\Delta) = N_e G(T_e) \{ [O_2] - [O_2(a^1\Delta_g)] g_1 g_0^{-1} \exp(E_1 T_e^{-1}) \}, \quad (13)$$

$$L(\Sigma) = N_e H(T_e) \{ [O_2] - [O_2(b^1\Sigma_g^+)] g_2 g_0^{-1} \exp(E_2 T_e^{-1}) \}, \quad (14)$$

where $G(T_e) = 1.143 \cdot 10^{-14} \exp[f_1(1/1500 - 1/T_e)]$, $H(T_e) = 1.616 \cdot 10^{-16} \exp[f_2(1/1500 - 1/T_e)]$, $E_1 = 11338$ K (or 0.977 eV) and $E_2 = 18881$ K (or 1.627 eV) are the energies of the $a^1\Delta_g$ and $b^1\Sigma_g^+$ levels of O₂ (Itikawa *et al.*, 1989), $f_1 = \{13200 + 1410 \sin[0.000241 (T_e - 500)]\} \{1 + \exp[(T_e - 14011)/1048]\}$, $f_2 = \{19225 + 560 \sin[0.000383(T_e - 1000)]\} \{1 + \exp[(T_e - 16382)/1760]\}$, the units of T_e are K and the units of $G(T_e)$ and $H(T_e)$ are $eV cm^3 s^{-1}$, $g_1/g_0 = 5/3$ and $g_2/g_0 = 1/3$ (Radzig and Smirnov, 1980).

Taking into account the results of observations and calculations of O₂($a^1\Delta_g$) and O₂($b^1\Sigma_g^+$) emissions and the production and loss rates of O₂($a^1\Delta_g$) and O₂($b^1\Sigma_g^+$) in chemical reactions given by Lopez-Gonzalez *et al.* (1989) and Yee *et al.* (1997) and using the MSIS-86 neutral model (Hedin, 1987) we found that $[O_2]$ is larger than $[O_2(a^1\Delta_g)] g_1 g_0^{-1} \exp(E_1 T_e^{-1})$ or $[O_2(b^1\Sigma_g^+)] g_2 g_0^{-1} \exp(E_2 T_e^{-1})$ at the E and F region altitudes of the ionosphere. As a result, we can compare

$Q(T_e)$ with $G(T_e)$ and $H(T_e)$ to evaluate the effects of $L(\Delta)$ and $L(\Sigma)$ on T_e in comparison with the effects of $L_{vib}(O_2)$ on T_e . The resulting values of the ratios $G(T_e)/Q(T_e)$ and $H(T_e)/Q(T_e)$ are less than $3.2 \cdot 10^{-4}$ and $4.5 \cdot 10^{-6}$ for $T_e \leq 1500$ K. In the electron temperature range 1500–3000 K, we have found that these ratios are less than 0.012 and 0.001. Thus, we can conclude that the value of $L_{vib}(O_2)$ is much larger than the value of $L(\Delta)$ or $L(\Sigma)$ and the effects of $L(\Delta)$ and $L(\Sigma)$ on the calculation of T_e are negligible.

4 The electron cooling rate by rotational excitation of O₂

The general expression for the rate of electron energy loss associated with rotational transitions in O₂($v = 0$) is given by Eq. (1) of Stubbe and Varnum (1972). Using the quadrupole Born approximation for rotational excitation cross sections given by Gerjuoy and Stein (1955) and making use of the fact that the rotational states are so close together that the summation can be changed into an integration (the quadrupole Born approximation formula of Mentzoni and Row (1963)), the electron cooling rate by means of rotational excitation of O₂($v = 0$) can be calculated as

$$L_{rot}(O_2(v = 0)) = C [O_2(v = 0)] N_e (T_e - T_n) T_e^{-0.5}, \quad (15)$$

where $C = (64\pi B a_0^2 Q^2 2^{0.5}) / (15\pi^{0.5} m_e^{0.5})$, B is the rotational constant for the ground vibrational state of O₂, Q is the quadrupole moment averaged over the ground vibrational state of O₂, and a_0 is the Bohr radius.

Dalgarno *et al.* (1968) used the value of an effective quadrupole moment given by Hake and Phelps (1967) as $Q = -1.8ea_0^2$ where e is the electronic charge. This value of Q was chosen by Hake and Phelps (1967) to obtain agreement between the calculated and experimental drift velocities of electrons. As a result, Dalgarno *et al.* (1968) found $C = 7 \cdot 10^{-14} eV cm^3 s^{-1} K^{-0.5}$ (the rotational constant $B = 1.793 \cdot 10^{-4} eV$ given by Radzig and Smirnov (1980) and $Q = -1.8ea_0^2$ give more accurate value of $C = 6.8 \cdot 10^{-14} eV cm^3 s^{-1} K^{-0.5}$). However, Lawton and Phelps (1978) assumed later that $Q = -0.3ea_0^2$ in their analysis of calculated and measured electron transport coefficients. The value of C is proportional to Q^2 so that the rate of electron energy loss associated with rotational transitions in O₂ given by Dalgarno *et al.* (1968) should be decreased by a factor of 36.

The cooling rate of electrons in O₂ was measured by Smith and Dean (1975) over the electron temperature range 350–800 K. They found that their results are well explained if the quadrupole Born approximation for rotational excitation cross sections with a value of $Q = -0.5ea_0^2$ is used. This result is the most practical for us since just the cooling rate of electrons in O₂ is studied in this work. As a result of the choice of Q given by Smith and Dean (1975), we can find that $C = 5.2 \cdot 10^{-15} eV cm^3 s^{-1} K^{-0.5}$.

The total rate of electron energy loss associated with rotational transitions within each vibrational level of

$O_2 L_{rot}(O_2) = \sum_{v=0}^{\infty} L_{rot}(O_2(v))$. To simplify the problem Dalgarno *et al.*(1968) used

$$L_{rot}(O_2) = C[O_2]N_e(T_e - T_n)T_e^{-0.5} . \quad (16)$$

This means that they did not distinguish between rotational excitation cross sections of O₂($v=0$) and O₂($v>0$). The number density of vibrationally unexcited O₂ is much larger than the number density of vibrationally excited O₂ during the undisturbed and geomagnetic storm periods at middle latitudes (Pavlov, 1989, 1994, 1998). As a result, the collisions of thermal electrons with vibrationally unexcited O₂ yield the main contribution for the rate of electron energy loss associated with rotational transitions within O₂($v=0$) and O₂($v>0$) and the possible error of Eq. (16) is expected to be small.

Finally, the electron cooling rate due to rotational excitation of O₂ by electron impact found in this study is less than that given by Dalgarno *et al.*(1968) by a factor of 13.

5 Conclusions

The electron energy transfer rates for vibrational excitation of O₂ have been calculated and fitted to analytical expressions by the use of the revised vibrationally excited O₂ cross sections (Allan, 1995). We found that within the electron temperature range 300–6000 K. the new energy transfer rates and the electron cooling rate obtained by using the Allan (1995) vibrational cross sections are much larger than those of Prasad and Furman (1973) obtained by using the Linder and Schmidt (1971) data. We have shown that the relative contribution of the vibrational levels $v=5, 6, 7$ to the electron cooling rate and the production frequency of the O₂ vibrational quanta is less than 0.5% for $T_e \leq 6000$ K and the vibrational levels $v=1-4$ are enough to calculate L and W .

This ²Π_g resonance dominates the vibrational cross sections in O₂ at energies below about 2.5 eV while the ⁴Σ_u⁻ and ²Σ_u⁻ resonances of O₂ give the major contribution to the O₂ vibrational cross sections above 3–4 eV. We found that the effects of the ⁴Σ_u⁻ and ²Σ_u⁻ resonances are negligible on the calculation of Q_{0v} , L , and W in comparison with those for the ²Π_g resonance. The results of this study also provide sufficient evidence to neglect the effects of the excitation of the $a^1\Delta_g$ and $b^1\Sigma_g^+$ electronic states of O₂ with thermal electrons on the calculation of T_e .

In this study, our goal was also to evaluate the accuracy of the electron cooling rate due to rotational excitation of O₂ found by Dalgarno *et al.* (1968). As we have shown, this currently accepted rate of electron energy loss associated with rotational transitions in O₂ must be decreased by a factor of 13.

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